

Expanded graphite / Polyaniline / F & O4 Composite

Preparation and wave absorbing property

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Abstract: expanded graphite was prepared by in-situ polymerization method / Polyaniline (Eg/pani Composites , will Fe₃O₄ is loaded on Eg/pani Surface ,

to obtain a with electromagnetic absorption properties Eg/pani/Fe₃O₄ Composite . by scanning the electron microscope (SEM) , X-ray diffraction (XRD) ,

Fourier Transform infrared spectrometer (FTIR) and Vector Network Analyzer (VNA the appearance of composite materials , the component and absorbing properties are studied investigate . The Wave-absorbing performance analysis results show that , When the doping concentration is 0.05 mol/l , Match thickness d = 2 mm , Minimum reflection loss for samples (RL

in 8.5 GHz at the - Panax DB. increases with doping concentration , Minimum reflection loss peak moves to low frequency , The corresponding thickness of the is gradually thickened . Dielectric Relaxation polarization of the material , Eddy Current loss and A/4 Interference cancellation of the model Shuangfeng , makes the Eg/pani/Fe₃O₄ Composite in electrical The field of magnetic wave absorption has certain application prospects .

Keywords: expanded graphite ; Conductive Polymers ; Magnetic Material ; Composite ; Wave-absorbing performance diagram category number O633.5; TM25 document marker A

with the wide application of electronic products , Electromagnetic Interference , electromagnetic pollution and electromagnetic leaks are getting worse , prepare for civilian and military purposes subject to concern . and a absorbing material is a functional material that is consumed by converting electromagnetic waves projected onto its surface into heat or other energy. . Composite absorbing material can effectively improve the electromagnetic environment , reduce electromagnetic leaks , is a new functional material ^,.

ferrite as a traditional nano-absorbing material , has super paramagnetic and nano effects , absorbs performance strongly , The has in the field of electromagnetic absorption Wider application ⁰ . But its disadvantages such as high density and easy oxidation limit its application ^H . Polyaniline (PANI) is a good stability , raw Material cheap , synthetic Easy , conjugated conductive polymers for high temperature and antioxidant properties , The has in the field of electromagnetic absorption _ The applied price set by value ^S . The composites of both have a great concern because of their electrical losses and magnetic losses . Hou etc ¹⁶ using hard template method to prepare hollow structure F & O₄/pani Minimum reflection loss for microspheres (RLm in) on GHz at -24.3 DB . Guo Yajun ¹⁷ The prepared the polyaniline / F & O₄ / Carbon black composites , to C band (4~8 GHz) and Ku band (12~18 GHz) has better absorption of electromagnetic waves . and carbon black than , worm-shaped expanded graphite (EG) as a new type of carbon absorber , due

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to porous network holes , to have a larger area than the table , to produce a high activity fresh surface , and an increase

in the interface layer will generate multiple layers of reflection on electromagnetic waves or Absorbs the, and to Increase the wave-absorbing ability. Zhao et al.⁸ prepared EG/Fe₃O₄ Nano-ring, its μ in 6.8 GHz at -24.8 DB. Chen, and so on⁹ prepared EG/PANI/Fe₃O₄ Composite, its RL in 13.28 GHz at -19. DB. for this, will Fe₃O₄ load on EG/PANI matrix, Get loose porous structure ternary composite material, is expected to improve the electromagnetic matching effect of materials, from the To improve its wave absorbing performance.

This article uses the in-situ growth method to prepare the EG/PANI Composite, and Fe₃O₄ to EG/PANI Composite for adsorption, gets the with both electrical loss performance and magnetic loss performance. EG/PANI/Fe₃O₄ composite, and study different HCl Doping to EG/PANI/Fe³⁺ Effects of absorbing properties of composite materials, prepared by EG/PANI/Fe³⁺ Composite has a certain wave-absorbing performance, good wave performance.

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1. Experiment Section

1.1 Reagents and instruments

aniline (An), Parse Pure, Tianjin da Mao chemical reagent factory, use after decompression distillation; ammonium persulfate (APS), Parse Pure, Tianjin City Heng Xing Chemical Reagent Manufacturing Co., Ltd. seven hydrated ferrous sulfate (FeSO₄·7H₂O), Parse Pure, Tianjin da Mao chemical reagent factory; six hydrate ferric chloride (FeCl₃·6H₂O), Parse Pure, Yantai Chemical Co., Ltd.; Ammonia (Quality Score 25%~28%, parsing Pure, Silver Companion Chemical Reagent Co., Ltd.; Hydrochloric acid (HCl, Quality Score 36%~38%, Parse Pure, Silver Companion chemical Reagent Limited companies; anhydrous ethanol, Parse Pure, Tianjin Fu Yu Fine Chemical Co., Ltd.; expanded graphite (EG), Excellent-level pure, Qingdao Tian Yuan da graphite Co., Ltd.; two times distilled water for homemade.

S-4800 Type field emission scanning electron microscope (SEM); Japan Hitachi Corporation; D8/advance type X-ray diffraction (XRD) instrument, German Brooks company; NEXUS-670 type Fourier transform IR spectrum (FTIR) instrument, USA Hi-Power company; PNA. 5244 A type vector Amount Network analyzer, agilent.

EG/PANI Preparation of composite materials

references [A] method to prepare the EG/PANI Composite. will 0.1 g EG to join 2 mL The ultrasonic dispersion in distilled water is complete all, join 0.40 mol/L Hydrochloric acid and 0.50 mL aniline, Ultrasonic Scatter complete. Transfer the system to three bottles, will again contain the 1. g APS 100 mL Water Solution constant pressure 0.5 h Add to above solution, Magnetic Stirring 1 h, centrifugal Drying, To get EG/PANI composites.

Keep the amount of expanded graphite constant, The amount of other reagents is increased proportionally to (Table 1), Repeat the above experiment as a control experiment.

EG/PANI/Fe₃O₄ Preparation of composite materials

references [B] method to prepare the Fe₃O₄ Magnetic nanoparticles. adds a certain amount of FeSO₄·7H₂O and FeCl₃·6H₂O mass ratio for 1:2 blending, and dissolving it to 100 mL in distilled water ultrasonic disperse evenly. Transfer the mixture to three bottles, under protection Water bath heating to 80 °C. Add ammonia reaction 2 h after, heats up to 100 °C And keep the warm 1 h. washed, gets after drying Fe₃O₄ Magnetic Nano granule.

will 4 g ANI-3 Composite and 0.1 g Fe₃O₄ magnetic nanoparticles added to 10 mL anhydrous Ethanol, Ultrasonic Dispersion completely, transfer to three-port bottle mechanical stirring 3 h, Magnetic Separation, Dry, gets the EG/PANI/Fe₃O₄ Composites.

2. Results and discussions

2.1 Sample Characterization

Diagram 1 Scanning electron microscope photos for different materials. From Diagram 1 (A) to see, EG renders a loosely porous structure, with a good slice layer structure. The that is prepared in this article EG Good expansion, has a fresh, active surface, is PANI. The in-situ growth of provides the basis for. Diagram 1 (B) ~ (E) is different PANI Packet-covered EG / PANI SEM photos of composite materials. Diagram 1 (B) The amount of aniline is 0.50 mL. EG / PANI Composite Shape, to see, PANI in the EG Block reunion on, and accompanies PANI Fiber-like aggregates. Diagram 1 (C), when the amount of aniline is reduced to 0.25 mL, PANI The package for is larger than the figure 1 (A) is noticeably smaller, but still has PANI particle reunion in EG Surface. Further reduce the amount of aniline to 0. ML, from Diagram 1 (D) to see the clearly PANI Evenly to overwrite EG layer Structure, with better wrapping effect. Also, from Diagram 1 (G) EDS The spectrogram can see, There are in composite materials N, Cl and Selement, where Cl The element originates from the HCl doped, S The element comes from an oxidizer ammonium persulfate, proved to be covered in expanded graphite surface particles to PANI^M. When the amount of aniline is lowered to the 0. ten mL when, from Diagram 1 (E) to see, PANI still enable EG End full cover, But the amount of coating is significantly higher than that of aniline 0. ML less. In the EG / PANI Composite Load Fe₃O₄ Experiment, uses aniline for 0. ML.

Diagram 4 gives the EG, PANI, Fe₃O₄, EG / PANI and EG / PANI/Fe₃O₄ Composite TGA Curve. When the temperature reaches All °C, EG and Fe₃O₄ The has a lower rate of weightlessness, is 8.3% and 8.5, and PANI The weight loss rate for is up to 17%. PANI weightlessness of can be attributed to 3 phases: adsorption of water loss and reaction of aniline monomer volatilization, The volatilization of low polymers and the decomposition of polymer molecular chains. EG / PANI Composite in All C The weightlessness rate for is 28%, to calculate the in the composite PANI The mass fraction of is approximately 45.4%, is lower than the composite in PANI The theoretical content of, This is because the PANI and EG There is an interaction between, Enhanced composite material Material Thermal stability⁰²². Ternary EG / PANI/Fe₃O₄ The final content of the composite is 81.0%, to calculate the in the composite Fe₃O₄ has a mass score of +1, slightly Lower Fe₃O₄ Add amount 50%.

2.2 Electromagnetic Wave absorption performance analysis

takes into account the HCl Doping amount to PANI Electrical Performance Impact, with 0, 0.1, 0.2 and 0.3 mol/L for HCl to PANI to mix Miscellaneous, get corresponding EG / PANI/Fe₃O₄ Composites, samples are recorded as S1, S2, S3 and S4. combine composites with paraffin wax by quality more than 1:1 Mix, and then suppress the mixed sample to the inside diameter 3. mm, outside diameter is 7. mm, thickness is 2. mm coaxial ring, on 2~18 GHz The frequency range tests its absorption performance for electromagnetic waves.

Diagram 5 gives the 2~18 GHz Range Samples S1, S2, S3 and S4 The electromagnetic parameters of the. From Diagram 5 (A) to see, Sample S1's Duplicate permittivity Real part U') The value is much larger than the other 3 Sample, from The decrements to 5. with HCl An increase in the amount of doping, $\hat{\epsilon}'$ value decreases, and each sample, values decrease as frequency increases, S2, S3 and S4 lower rate of is smaller. same, in **Figure 5** (B), Sample S1 for complex permittivity imaginary part U'') The value is also larger than the other 3 Sample, from the 5 Decrease to 6 With a larger trend, S2, S3 and S4 for, The value is 9~5.46, 5.2~3.1 and 4.2~3.1, shows a trend of decreasing first and then increasing for the entire frequency range. Diagram 5 (C), - like product S1 Complex permeability of (/T) The change in is significantly different from the other 3 Sample, in 2~10 and 15~18 GHz Is in a downward trend, in 10~15 GHz The scope is an increasing trend. and sample S2, S3 and S4 for $\hat{\mu}'$ The value is in the 6.5 and ten. 5GHz minimum appears near is the same as the maximum and the trend is similar. Diagram 5 (D) shows the complex permeability of the sample ($\hat{\mu}''$) with frequency change relationship. Sample S2, S3 and S4 a downward trend throughout the frequency range, and sample S1 decreases before increasing, in GHz minimum value near, and 4 Sample $\hat{\mu}''$ values have varying degrees of negative. A positive part of the value is used to reflect the loss of magnetic energy. Conversely, $\hat{\mu}''$ The value is a negative representation of the magnetic energy from EG/ PANI / Fe⁺ radiation out of composites. The energy emitted by the may be caused by the motion of the charge under alternating. Root According to Maxwell's equation, The carrier in an alternating magnetic field forms a vortex current under the influence of a inductive electric field, and produce an opposite inductive power Farm, convert partial electric field to

magnetic field to radiate from inside material, cause value to drop negative⁰²³²⁴.

based on the transmission line theory, reflection loss of a single layer absorbing coating at the time of vertical incidence of electromagnetic waves (RL) can be emulated with the following model

Diagram 6 gives the EG / PANI / Fe₃O₄ Reflective loss curves for different thicknesses of composites. Contrast Chart 6 to see, when the matching thickness is 1 mm, Sample S1 Minimum reflection loss for (RL_{min}) on, GHz at -9.4 DB; Sample S2 in D = 2 mm in 8.5 GHz Place RL_{min} reaches -37 DB, corresponds to RL < -10 DB The frequency width of the is 1. GHz (7.92~9. GHz); Sample S3 at the matching thickness d = 4 mm on 4.4 GHz Place RL_{min} reached -30.8 DB, Bandwidth is 1. GHz (4.24~5. GHz); same, Sample S4 on Match thickness d = 4.5 mm on 4.24 GHz Place RL_{min} reach -40 DB, bandwidth to 0.96 GHz (3.76~7.72 GHz). contrast 4 Sample Discovery, with HC1 An increase in the amount of doping, the minimum reflection loss peaks of the sample are gradually moving to the low-frequency. Considering the reflection loss of the sample consumption and matching thickness, This article with S2 as example, comparison 4 sample RL < -10 dB Bandwidth. at the matching thickness D = 1.5 mm, Sample S1, S2, S3 and The bandwidth for the S4 corresponds to 0.224~64, GHz, 1.92 (+ 4 GHz ") and 2. (12.72~14.8GHz). to discover, by adjusting the HCl The doping amount of, to make the EG / PANI / Fe₃O₄ Composite to different frequencies The electromagnetic waves of the section are absorbed in varying degrees.

When the electromagnetic wave hits the surface of the material, will not only encounter reflection on the surface of the absorbing material, and will also encounter an anti-on the metal substrate fire, on the 2 The same frequency of reflection waves, if 2 Wave path difference for reflection wave (4) for odd multiples of half wavelength will appear dry

When the electromagnetic wave enters the absorbing material, its wavelength A the represents the with the following expression:

with S2 is the research object, diagram 7 the gives its minimum counter fire-loss peaks corresponding to A/4 Model. to see, when match thickness 3.0, 3.5, 4.0, 4.5 and 5.0 mm when,

A appears under the same matching thickness 2 feature absorption peaks (divide To be 5., 6 GHz 4.72, 15 GHz;

4., a GHz; 3.52, 11.44 GHz; 3.,

10.32 GHz), and with increased matching thickness, Shuangfeng The band width between decreases gradually. Description EG / PANI / Fe₃O₄ The electromagnetic absorber at the minimum reflection loss peak of the composite material A/4 and 3 A/4 Model. The main reason for is that the absorbing material The reflection waves on the surface and the reflection waves of the matrix are eliminated. indicates that interference is eliminated in EG / PANI / Fe₃O₄ Composites The Electromagnetic absorption process of the plays a significant role in.

Electrical loss tangent (ϵ''/ϵ') and magnetic loss tangent (μ''/μ') the reflects the electrical loss energy of the material to the electromagnetic wave Force and magnetic loss Ability⁰²⁹. from Diagram 8 to see, with EG / PANI / Fe₃O₄ Composites HCl increases the amount of doping, The electrical of the material The loss tangent values are decreasing, and their magnetic loss tangent values are increasing. and the electrical loss tangent value (>0.) throughout the frequency The rate range is greater than the magnetic loss angle tangent (<0), description power loss in EG / PANI / Fe₃O₄ composites absorbing electromagnetic waves The contribution of the process is greater than the loss of the magnet. The reason for the electrical loss caused by IS: (1) EG / PANI Electronic jumps on the surface, To form a conductance^{B0'} to; (2) A conductive network may be formed inside a composite material, causing electrical loss^M.

to study dielectric loss and magnetic loss of materials, Many scholars introduce Cole ^ Cole semicircle and Co ^ ^ k ^)⁻². Corpse ^ theory ^ Sample S1 as an example, by Diagram 9 (A) can be seen in the, is in the 30~45 Scope, ϵ'' is in the 14~23 in the range, EG / pani/fe₃O₄ The Composite has a cole ^ cole semicircle. because of its, and, value decreases with frequency increase, can be inferred from the 10~18 GHz appears in the scope of cole ^ cole semicircle, The indicates that the material has dielectric relaxation in the High-frequency range, The is primarily made up of dipole polarization and interface poles cause ^ 38. is usually, Magnetic losses are mainly derived from domain wall

resonances, Eddy Current loss, Natural resonance and exchange resonance On, and Domain wall resonance on 2~18 GHz Band is negligible. The Magnetic loss of the sample may be caused by an exchange resonance, This is because the F & O4 The ruler of the nanoparticles in ten nm around, Small particle size, is advantageous for resonance exchange behavior^[1]. from Diagram 9 (B) To See, for the material C0 value is gradually with frequency Decrease, on />8 GHz Basic remain constant, This description EG/pani/ Fe³⁺O₄ The absorption of electromagnetic waves by composite materials in the /<8 GHz Natural resonance in range, />8 GHz within range of Eddy current loss main.

3. knot theory

the uses the in-place aggregation method to prepare the Eg/pani composite, and Fe₃O₄ to eg/pani composite material adsorbed, gets the with electromagnetic performance EG/pani/Fe₃O₄ Composite, and pass HC1 to modify the. Research Discovery, Low Doping concentration Benefits The material's absorption of electromagnetic waves. When the doping concentration is 0. mol/l, Match thickness D = 2 mm, in 8. \$ GHz at RL_{min} to reach -37 DB. increases with doping concentration, The frequency of the minimum reflection loss peaks of the sample decreases when it appears (64~4.24 GHz), match thickness gradually thicken (1~4 5 mm). by adjusting the HCl Doping concentration, to effectively adjust the frequency bands of the best absorption peaks. The results of the composite electromagnetic parameters indicate that, electrical losses contribute significantly to the electromagnetic absorption process. The material has a dielectric in the High-frequency range yu phenomenon, and 2~8 GHz The magnetic losses of materials within the range are mainly natural resonances, in 8~18 GHz Eddy Current loss in scope is the primary. above loss and A /4 Shuangfeng of interference cancellation in the model, makes the EG / pani/Fe₃O₄ Composite Pairs in 2~18 GHz Scope Partial band electromagnetic shielding within has _ Application Prospects for _.

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